Synthesis of syn-Substituted Triptycenes via Heteroatom-directed Metallation

Takeshi Kawase, Nobuo Asai, Tatsuo Ogawa, and Masaji Oda*

Department of Chemistry, Faculty of Science, Osaka University, Toyonaka, Osaka 560, Japan

Heteroatoms in 9-substituted triptycenes direct *syn*-lithiation at the benzene rings, providing a new method for the synthesis of 1,9-di- and 1,8,9-tri-substituted triptycenes.

syn-Substituted triptycenes with, e.g., 1,8,13-tri- and 1,8,9,13tetra-substitution, are of interest in terms of spatial interactions between the substituents, which are closely spaced with parallel bonding to the rigid carbon framework of triptycene, in host-guest chemistry, etc. Although several of these compounds have been synthesised via benzyne-anthracene cycloaddition reactions, as yet no selective synthesis has been described. $^{\rm 1}$

In view of the regioselective *ortho*-lithiation of aromatic compounds assisted by heteroatoms,² it is expected that an appropriate heteroatom functionality in the substituent at the bridgehead (C-9 and C-10) of triptycene might direct, in



(1) X = Y = H(2) X = H, Y = CHO(4) X = OMe, Y = H(5) X = OMe, Y = CHO(7) $X = CH_2OMe, Y = H$ (8) $X = CH_2OMe, Y = CHO$ (9) X = CONHMe, Y = H(12) a; X = CONHMe, Y = OHb; X = CONHMe, Y = CH(OH)Phc; X = CONHMe, Y = Br

principle, up to three syn- metallations at C-1, C-8, and C-13, by taking advantage of the D_{3h} symmetry and nearly independent nature of the three benzene rings of triptycene. We report here the selective synthesis of 1,9-di- and 1,8,9-trisubstituted triptycenes by heteroatom directed syn-lithiation.

Triptycene (1) itself was lithiated only slowly and nonselectively by treatment with the base complex BunLi-TMEDA (TMEDA = tetramethylethylenediamine) (1:1; 2)equiv., ether, N₂, room temp., 12 h) followed by addition of excess dimethylformamide (DMF) at 0 °C, to afford 1- and 2-formyltriptycene (2) and (3)³ in 21 and 8% yields, respectively, with 69% recovery of (1). Under similar conditions, the lithiation of 9-methoxytriptycene (4) proceeded more smoothly and preferentially at C-1, to give, on formylation, 1-formyl-9-methoxytriptycene (5) and 1-formyl-10-methoxytriptycene (6)[†] in 47 and 7% yields [28% recovery of (4)]. 9-Methoxymethyltriptycene (7), in which the oxygen atom can approach C-1, C-8, and C-13 more closely than that of (4), was lithiated even more effectively and selectively, leading to 1-formyl-9-methoxymethyltriptycene (8) in 70% yield [15% recovery of (7)]. Use of 3 equivalents of the base increased the vield of (8) to 86% after 5 h of lithiation. However, no 1,8-diformyl compounds were formed by syn-1,8-dilithiation with these ethers, although some 1,5-diformyl compounds were isolated in poor yields under more forcing conditions.

The syn-dilithiation and the highly efficient monolithiation were substantiated using the bidentate and highly directive sec-carboxamide.⁴ Thus, lithiation of N-methylamide (9) with the base (3 equiv.) at 0 °C for 15 min gave, after formylation, the hydroxylactam (10) in 88% yield; a similar reaction using 4 equivalents of the base at 25 °C for 40 min gave the formyl lactam (11) (a mixture of two possible stereoisomers) in 60% yield in addition to (10) (32%). The dilithiated intermediate is (14). Other bidentate functional groups such as acetal and oxazoline, however, were ineffective, resulting mostly in decomposition or recovery of the starting materials.



(14)

These syn-lithiations allow the synthesis of a variety of 1,9-di- and 1,8,9-tri-substituted triptycenes, in particular functionalised ones; for example, (12a-c) were obtained in good yields, via monolithiation of (9) upon quenching with oxygen (60%), benzaldehyde (88%), and 1,2-dibromoethane (67%), respectively. The dilithiation and oxygen-quenching of (9) afforded diphenol-amide (13) (52%) together with (12a) (32%).

To conclude, *syn*-substituted triptycenes are accessible through heteroatom-directed metallation. Suitable tridentate heteroatom functionality at C-9 position would enable *syn*-trimetallation and -functionalisation of triptycenes.

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References

- S. Akiyama, S. Misumi, and M. Nakagawa, *Bull. Chem. Soc. Jpn.*, 1960, **33**, 1293; G. Yamamoto and M. Oki, *Bull. Chem. Soc. Jpn.*, 1984, **57**, 2219; M. E. Rogers and B. A. Averill, *J. Org. Chem.*, 1986, **51**, 3308; Y. Nakai, G. Yamamoto, and M. Oki, *Chem. Lett.*, 1987, 89.
- 2 For reviews see: H. W. Gschwend and H. R. Rodriguez, Org. React., 1979, 26, 1; P. Beak and V. Snieckus, Acc. Chem. Res., 1982, 15, 306.
- 3 S. Tivakornpannarai and E. E. Waali, J. Am. Chem. Soc., 1986, 108, 6058.
- 4 D. W. Slocum and C. A. Jennings, J. Org. Chem., 1976, 41, 3653.

[†] Satisfactory analytical and spectral data were obtained for all new compounds. The structures of (5) and (6) were determined from their ¹H NMR spectra, in comparison with those of (2) and (3), particularly the splitting of the formylated benzene ring protons, the chemical shifts of the bridgehead protons and formyl protons. For (5): m.p. 224 °C, ¹H NMR (CDCl₃, 500 MHz) δ 4.28 (3H, s), 5.39 (1H, s, H-10), 7.00 (1H, t, J7.6 Hz), 7.09—7.17 (4H, m), 7.44 (1H, d, J7.4 Hz), 7.47 (2H, d, J7.4 Hz), 7.54 (1H, d, J7.8 Hz), 7.69 (2H, d, J7.1 Hz), 11.39 (1H, s, CHO) (long range coupling omitted). For (6): m. p. 196 °C, ¹H NMR (CDCl₃, 100 MHz) δ 4.33 (3H, s), 6.76 (1H, s, H-9), 6.96—7.50 (8H, m), 7.55—7.70 (2H, m), 7.82 (1H, d, J7.5 Hz, H-2), 10.23 (1H, s, CHO).