## **Synthesis of syn-Substituted Triptycenes** *via* **Heteroatom-directed Metallation**

## **Takeshi Kawase, Nobuo Asai, Tatsuo Ogawa, and Masaji Oda\***

Department of Chemistry, Faculty of Science, Osaka University, To yonaka, Osaka **560,** Japan

Heteroatoms in 9-substituted triptycenes direct syn-lithiation at the benzene rings, providing a new method for the synthesis of 1,9-di- and 1,8,9-tri-su bstituted triptycenes.

syn-Substituted triptycenes with, *e.g.,* 1,8,13-tri- and 1,8,9,13 tetra-substitution, are of interest in terms of spatial interactions between the substituents, which are closely spaced with parallel bonding to the rigid carbon framework of triptycene, in host-guest chemistry, etc. Although several of these compounds have been synthesised *via* benzyne-anthracene cycloaddition reactions, as yet no selective synthesis has been described.<sup>1</sup>

In view of the regioselective orrho-lithiation of aromatic compounds assisted by heteroatoms,<sup>2</sup> it is expected that an appropriate heteroatom functionality in the substituent at the bridgehead (C-9 and C-10) of triptycene might direct, in



 $(2)$  **X** = **H**, **Y** = **CHO**  $(4)$   $X =$  OMe,  $Y = H$ *(5)* **X** = OMe, Y = CHO  $(7)$   $X = CH_2OMe, Y = H$  $(8)$  X = CH<sub>2</sub>OMe, Y = CHO  $(9)$  X = CONHMe, Y = H  $(12)$  **a**;  $X = \text{CONHMe}$ ,  $Y = \text{OH}$  $\mathbf{b}$ ;  $\mathbf{X} = \text{CONHMe}$ ,  $\mathbf{Y} = \text{CH}(\text{OH})\text{Ph}$  $c; X = \text{CONHMe}, Y = \text{Br}$ 

principle, up to three syn- metallations at C-1, C-8, and C-13, by taking advantage of the *D3h* symmetry and nearly independent nature of the three benzene rings of triptycene. We report here the selective synthesis of 1,9-di- and 1,8,9-trisubstituted triptycenes by heteroatom directed syn-lithiation,

Triptycene **(1)** itself was lithiated only slowly and nonselectively by treatment with the base complex BunLi-TMEDA (TMEDA = tetramethylethylenediamine)  $(1:1; 2)$ equiv., ether,  $N_2$ , room temp., 12 h) followed by addition of excess dimethylformamide (DMF) at  $0^{\circ}$ C, to afford 1- and 2-formyltriptycene **(2)** and **(3)3** in 21 and 8% yields, respectively, with 69% recovery of **(1).** Under similar conditions, the lithiation of 9-methoxytriptycene **(4)** proceeded more smoothly and preferentially at C-1, to give, on formylation, 1-formyl-9-methoxytriptycene *(5)* and 1-formyl-10-methoxytriptycene **(6)t** in 47 and 7% yields [28% recovery of **(4)].**  9-Methoxymethyltriptycene **(7),** in which the oxygen atom can approach C-1, C-8, and C-13 more closely than that of **(4),** was lithiated even more effectively and selectively, leading to **1-formyl-9-methoxymethyltriptycene (8)** in 70% yield [ 15% recovery of **(7)].** Use of 3 equivalents of the base increased the yield of **(8)** to 86% after 5 h of lithiation. However, no 1,8-diformyl compounds were formed by  $syn-1$ ,8-difithiation with these ethers, although some 1,5-diformyl compounds were isolated in poor yields under more forcing conditions.

The syn-dilithiation and the highly efficient monolithiation were substantiated using the bidentate and highly directive sec-carboxamide.4 Thus, lithiation of N-methylamide **(9)** with the base (3 equiv.) at  $0^{\circ}$ C for 15 min gave, after formylation, the hydroxylactam **(10)** in 88% yield; a similar reaction using 4 equivalents of the base at 25  $\degree$ C for 40 min gave the formyl lactam **(11)** (a mixture of two possible stereoisomers) in 60% yield in addition to **(10)** (32%). The dilithiated intermediate is **(14).** Other bidentate functional groups such as acetal and oxazoline, however, were ineffective, resulting mostly in decomposition or recovery of the starting materials.



These syn-lithiations allow the synthesis of a variety of 1,9-di- and 1,8,9-tri-substituted triptycenes, in particular functionalised ones; for example, **(12a-c)** were obtained in good yields, *via* monolithiation of **(9)** upon quenching with oxygen (60%), benzaldehyde *(%Yo),* and 1,2-dibromoethane  $(67\%)$ , respectively. The dilithiation and oxygen-quenching of **(9)** afforded diphenol-amide **(13)** (52%) together with **(12a)**   $(32\%)$ .

 $(14)$ 

To conclude, syn-substituted triptycenes are accessible through heteroatom-directed metallation. Suitable tridentate heteroatom functionality at C-9 position would enable syn-trimetallation and -functionalisation of triptycenes.

Received, *16th* October *1989; Corn. 9104461 B* 

## **References**

- 1 S. Akiyama, S. Misumi, and M. Nakagawa, *Bull. Chem. Soc. Jpn.*, 1960,33, 1293; *G.* Yamamoto and M. Oki, *Bull. Chem. SOC. Jpn.,*  1984, *57,* 2219; M. **E.** Rogers and **B. A.** Averill, *J. Org. Chem.,*  1986,51,3308; Y. Nakai, G. Yamamoto, and M. Oki, *Chem. Lett.,*  1987, 89.
- For reviews see: H. W. Gschwend and H. R. Rodriguez, *Org. React.,* 1979, *26,* 1; **P.** Beak and *V.* Snieckus, *Acc. Chem. Res.,*  1982, *15,* 306.
- **S.** Tivakornpannarai and E. E. Waali, *J. Am. Chem.* **SOC.,** 1986,  $\mathbf{a}$ *108,* 6058.
- D. W. Slocum and C. **A.** Jennings, *J. Org. Chem.,* 1976,41,3653.

t Satisfactory analytical and spectral data were obtained for all new compounds. The structures of *(5)* and **(6)** were determined from their <sup>1</sup>H NMR spectra, in comparison with those of  $(2)$  and  $(3)$ , particularly the splitting of the formylated benzene ring protons, the chemical shifts of the bridgehead protons and formyl protons. For *(5):* m.p. 224 **"C, lH** NMR (CDC13, **500** MHz) 6 4.28 (3H, s), 5.39 (lH, s, H-lo), **7.00(1H,t,J7.6Hz),7.09-7.17(4H,m),7.44(1H,d,J7.4Hz),7.47 (2H,d,J7.4Hz),7.54(1H,d,J7.8Hz),7.69(2H,d,J7.1Hz),** 11.39 (lH, s, CHO) (long range couplingomitted). For **(6):** m.p. 196"C, **1H**  (8H, m), 7.55-7.70 (2H, m), 7.82 (lH, d, *J* 7.5 Hz, H-2), 10.23 (lH, s, CHO). NMR (CDC13, 100 MHz) 6 4.33 (3H, **s),** 6.76 **(lH, S,** H-9), 6.96-7.50